

Apparent molar volumes and enthalpies of solution of tetracyanoethylene in some solvents and of butan-1-ol in n-octane at different concentrations

Kiselev V., Kashaeva E., Shihaab M., Kononov A.
Kazan Federal University, 420008, Kremlevskaya 18, Kazan, Russia

Abstract

The effects of specific solute-solvent and solute-solute interactions on the change in the apparent molar volumes and in the enthalpies of solution, were found to be proportional in the title solutions at 25 °C.

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